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Use with Chapter 7, Section 7.2 1. How many
valence electrons does a neutral magnesium
(Mg) atom have? 2. What is the charge on a
magnesium ion? What does magnesium have to do
to form such an ion, and why does it tend to
do so? 3. How many valence electrons does a
single neutral chlorine (Cl) atom have? 4.

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Matter and Change Chapter 5 . uest 11 Name:
Date: Hour: Information: Energy Levels and
Sublevels ... Considering your answer to
question 5, how many electrons can fit in the
entire 5t energy level? can (Avel is a ha
AC5a) 50đ 7. Write the notation for an
electron spinning clockwise in a p sublevel
...

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Series Use with Chapter 9, Section 9.2 1. For
each of the following pairs of elements,

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underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

TEACHING TRANSPARENCY WORKSHEET The Activity Series

Teaching Transparency Worksheets Chemistry Answers Chapter 19 Author:

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Teaching Transparency Worksheets Chemistry Answers Chapter 19

Teaching Transparency Worksheets Chemistry: Matter and Change 0 Chapter 8 . 1.7, the CF bond is polar covalent. (b) $\text{EN}_{\text{Cl}} - \text{EN}_{\text{P}} = 3.0 - 2.1 = 0.9$. Since $0.5 < \text{EN}_{\text{Cl}} - \text{EN}_{\text{P}} < 1.7$, the PCl bond is polar covalent. (c) $\text{EN}_{\text{Cl}} - \text{EN}_{\text{Li}}$ Section 8.5 Electronegativity and Polarity 265 Compare and contrast polar . that is in excess.

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