

## Student Exploration Chemical Equations Gizmo Answer Key

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To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the Reactants box and "H<sub>2</sub>O" in the Products box. This...

Student Exploration- Balancing Chemical Equations (ANSWER ...

Chemical Equations. Launch Gizmo. Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

Chemical Equations Gizmo : Lesson Info : ExploreLearning

Here are the recommendations: All adults - including people with various disabilities - should aim for at least 150 minutes of moderate aerobic activity, or 75 minutes of vigorous activity, or the equivalent combination of two during the week.

Student exploration balancing chemical equations gizmo ...

To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the Reactants box and "H<sub>2</sub>O" in the Products box. This...

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Student Exploration For Gizmo Answer Key Chemical Equations

Gizmo Warm-up. In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: H<sub>2</sub> + O<sub>2</sub> → H<sub>2</sub>O. How many atoms are in a hydrogen molecule (H<sub>2</sub>)? 2 How many atoms are in an oxygen molecule (O<sub>2</sub>)? 2

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Student Exploration: Chemical Equations Combustion – a fuel reacts with oxygen to release carbon dioxide, water, and heat. CH<sub>4</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O CH<sub>4</sub> + 2O<sub>2</sub> = CO<sub>2</sub> + 2H<sub>2</sub>O C<sub>3</sub>H<sub>8</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O C<sub>3</sub>H<sub>8</sub> + 5O<sub>2</sub> = 3CO<sub>2</sub> + 4H<sub>2</sub>O C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 6O<sub>2</sub> = 6CO<sub>2</sub> + 6H<sub>2</sub>O Single replacement – an element replaces another element in a compound. KCl + F<sub>2</sub> → KF + Cl<sub>2</sub> 2KCl + F<sub>2</sub> = 2KF + Cl<sub>2</sub> Mg + HCl → MgCl<sub>2</sub> + H<sub>2</sub> Mg + 2HCl = MgCl<sub>2</sub> + H<sub>2</sub> Cu + AgNO<sub>3</sub> → Cu(NO<sub>3</sub>)<sub>2</sub> + Ag Cu ...

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\_\_\_\_ Gizmo Warm-up In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: H<sub>2</sub> + O<sub>2</sub> → H<sub>2</sub>O. 1. How many atoms are in a hydrogen molecule (H<sub>2</sub>)? 2 2. How many atoms are in an oxygen molecule (O<sub>2</sub>)?

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2018 5. A chemical equation is balanced if the number of each type of atom on the left side is equal to the number of each type on the right side. Is this reaction balanced? \_\_\_\_ Activity A: Balancing equations Get the Gizmo ready: Check that the Synthesis reaction is selected and that all coefficients are set to one. (The coefficients are the numbers in the boxes.)

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Balancing Equations Gizmo - Student Exploration Balancing ... In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction:  $H_2 + O_2 \rightarrow H_2O$ . 1. Student Exploration: Balancing Chemical Equations

Student Exploration Balancing Chemical Equations Gizmo ...

To set up an equation in the Chemical Equations Gizmo™, type the chemical formulas into the text boxes of the Gizmo. First, type in “ $H_2+O_2$ ” in the Reactants box and “ $H_2O$ ” in the Products box. This represents the reaction of hydrogen and oxygen gas to form water. Student Exploration: Balancing Chemical Equations

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