

Section 11 Study Guide Stoichiometry Answer Key

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Stoichiometry Section 11.1 What is stoichiometry? In your textbook, read about stoichiometry and the balanced equation. For each statement below, write true or false. ____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.

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Stoichiometry STEP 1 Fold a sheet of paper in half lengthwise. STEP 2 Fold in half widthwise and then in half again. STEP 3 Unfold and the top flap to make four tabs. STEP 4 Label the tabs with the steps in stoichiometric calculations. &,\$!"%3Use this Foldable with Section 11.2. As you read this section, summarize each step on a tab and

Chapter 11: Stoichiometry

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Study Guide for Chapter 11 Stoichiometry 368 Chapter 11 • Stoichiometry Section 11.1.1 Objectives Describe the types of relationships indicated by a balanced chemical equation. State the mole ratios from a balanced chemical equation. Review Vocabulary reactant: the starting substance in a chemical reaction New Vocabulary

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In your textbook, read about stoichiometry and the balanced equation. For each statement below, write true or false. ____true____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry. ____true____ 2.

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