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Solution: A volume of 1 L equals 0.001 m³, see Table A.1.

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From Figure 2.7 the density is in the range of 10 so we get $m = 10 \cdot 0.001 \text{ m}^3 = 10 \text{ kg}$ A more accurate value from Table A.4 is 13 580 For the air we see in Figure 2.7 that density is about 1 so we get $m = 1 \cdot 0.001 \text{ m}^3 = 0.001 \text{ kg}$ A more accurate value from Table A.5 is 1.17 at 100 kPa, 25°C.

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relation to energy , work , radiation , and properties of matter. The behavior of these quantities is governed by the four laws of thermodynamics which convey a quantitative description using measurable macroscopic physical quantities , but may be explained in terms of microscopic constituents by statistical mechanics.

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