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Note the ratio of H^+ to OH^-

is 1 to 1 Step 2: Substitute

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values into the equation $M_A \times V_A = M_B \times V_B$
 $M_A(30. \text{ mL}) = (1.2 \text{ M})(50. \text{ mL})$ Step 3:
Solve for the unknown M 12M
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